# How to write a lab report



### **Purpose:**

- Lab reports are essential in all labs and represent the major part for grading your work.
- Specific protocols and skills are necessary to write a formal lab report and developments of these skills are essential for a successful laboratory experience.
- Witing lab reports are a good exercise for your senior thesis; a thesis has a very similar format!

# Prepare for the lab:

- Read the experiment from the lab manual.
- Understand the experiment to save time and avoid errors.
- Never use pencil in writing a lab report.

## **Organization:**

Sections should appear in your paper in the order described below. Each section must be typed in bold letters to differentiate it from the rest of the text.

- 1) Cover page
- 2) Introduction
- 3) Experimental (materials and methods)
- 4) Results
- 5) Discussion
- 6) Conclusions
- 8) References

### The lab report should include the following sections:

### A) Cover page:

It contains course name, course number, title of lab, course / lab sections, names of participants, name of teacher, and date.

### B) INTRODUCTION

Record objectives of the experiments, history, background of the problem, chemical reactions, and structures. This can be usually found in the lab manual.

#### C) MATERIALS AND METHODS

- Materials include equipments and reagents. Include actual concentrations, weights, volumes, and any precautions required.
- Methods include summary of procedures in your own words.
- Never copy the procedure from the lab manual
- Don't use "I"
- Don't use "First" or "Second" or "Third" etc.
- Don't use numbers to distinguish the steps but write complete sentence.

• Use the past tense and passive voice.

For example:

Not: Collect the precipitate in filter.

But: The precipitate was collected in a filter.

- Define all specialized terminology.
- Define abbreviations.

For example:

The Ethedium Bromide (EthBr) is a red color.

Place a space between a number and a unit such as 10 g.

• Do not start a sentence with a value.

For example:

Not: 2 gm of agarose powder was combined with 1X buffer solution.

But: Agarose powder (2 gm) was combined with 1X buffer solution.

### D) RESULTS:

Record tables, figures, calculations, and experiment results without discussing their implications. Hint: Do not start a sentence with "Figure 1" or "Table 1", etc.. Labelle the axes with the quantity divided by its unit.

### E) DISCUSSION:

- Discuss observations and the obtained results.
- Discuss what happened and why.
- Do not repeat the information from the procedure or results section.
- Avoid a lengthy discussion.

### F) CONCLUSIONS:

Write a short summary include:

- Record any new information.
- Record any mistakes and correct them.
- Record the information that you learned in the lab and how can apply it in to making a new idea and product.

#### G) REFRENCES

Record all references, such as journals or books.