

# 4-(4-Chlorophenyl)-8-methyl-2-oxo-1,2,5,6,7,8-hexahydroquinoline-3-carbonitrile

Abdullah M. Asiri,<sup>a,b</sup> Abdulrahman O. Al-Youbi,<sup>a</sup> Hassan M. Faidallah,<sup>a</sup> Khadija O. Badahdah<sup>a</sup> and Seik Weng Ng<sup>c,\*</sup>

<sup>b</sup>Chemistry Department, Faculty of Science, King Abdulaziz University, PO Box 80203 Jeddah, Saudi Arabia, <sup>a</sup>Center of Excellence for Advanced Materials Research, King Abdulaziz University, PO Box 80203 Jeddah, Saudi Arabia, and <sup>c</sup>Department of Chemistry, University of Malaya, 50603 Kuala Lumpur, Malaysia  
Correspondence e-mail: seikweng@um.edu.my

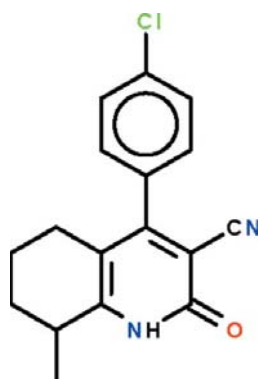
Received 5 September 2011; accepted 5 September 2011

Key indicators: single-crystal X-ray study;  $T = 100$  K; mean  $\sigma(\text{C}-\text{C}) = 0.004$  Å;  $R$  factor = 0.065;  $wR$  factor = 0.187; data-to-parameter ratio = 15.5.

The six-membered *N*-heterocyclic ring of the title compound,  $\text{C}_{17}\text{H}_{15}\text{ClN}_2\text{O}$ , is fused with a methyl-substituted cyclohexene ring. The approximately planar nitrogen-bearing ring (r.m.s. deviation 0.019 Å) is aromatic, and the N atom shows a trigonal-planar coordination; its benzene substituent is aligned at  $77.1(1)^\circ$ . The cyclohexene ring adopts a half-chair conformation. In the crystal, inversion-related molecules are linked by pairs of  $\text{N}-\text{H}\cdots\text{O}$  hydrogen bonds, generating dimers.

## Related literature

For a related compound, see: Asiri *et al.* (2011).



## Experimental

### Crystal data

$\text{C}_{17}\text{H}_{15}\text{ClN}_2\text{O}$   
 $M_r = 298.76$   
 Monoclinic,  $C2/c$   
 $a = 18.6304(4)$  Å  
 $b = 18.7399(4)$  Å  
 $c = 8.5209(2)$  Å  
 $\beta = 90.229(2)^\circ$   
 $V = 2974.89(11)$  Å<sup>3</sup>  
 $Z = 8$   
 Cu  $K\alpha$  radiation  
 $\mu = 2.27$  mm<sup>-1</sup>  
 $T = 100$  K  
 $0.30 \times 0.03 \times 0.03$  mm

### Data collection

Agilent SuperNova Dual diffractometer with Atlas detector  
 Absorption correction: multi-scan (*CrysAlis PRO*; Agilent, 2010)  
 $T_{\min} = 0.550$ ,  $T_{\max} = 0.935$   
 10387 measured reflections  
 3014 independent reflections  
 2682 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.025$

### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.065$   
 $wR(F^2) = 0.187$   
 $S = 1.03$   
 3014 reflections  
 194 parameters  
 H atoms treated by a mixture of independent and constrained refinement  
 $\Delta\rho_{\max} = 0.79$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.40$  e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{N1}-\text{H1}\cdots\text{O1}^i$	0.91 (4)	1.84 (4)	2.744 (3)	174 (4)

Symmetry code: (i)  $-x + 1, -y + 1, -z + 1$ .

Data collection: *CrysAlis PRO* (Agilent, 2010); cell refinement: *CrysAlis PRO*; data reduction: *CrysAlis PRO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *publCIF* (Westrip, 2010).

We thank King Abdulaziz University and the University of Malaya for supporting this study.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: XU5320).

## References

- Agilent (2010). *CrysAlis PRO*. Agilent Technologies, Yarnton, England.  
 Asiri, A. M., Faidallah, H. M., Al-Youbi, A. O., Alamry, K. A. & Ng, S. W. (2011). *Acta Cryst.* **E67**, o2468.  
 Barbour, L. J. (2001). *J. Supramol. Chem.* **1**, 189–191.  
 Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.  
 Westrip, S. P. (2010). *J. Appl. Cryst.* **43**, 920–925.