

Battery



- Source of pushing electrons
- Electrochemical reactions
- Look for types of batteries in Wikipedia!!

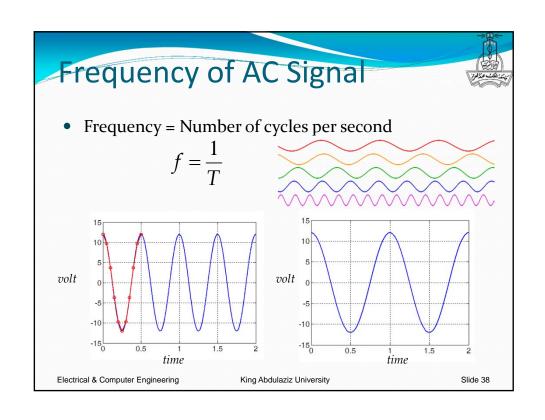
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• Cascade 73 batteries in series. Do they generate 110V? • DC: direct current • AC: alternating current volt volt time Electrical & Computer Engineering King Abdulaziz University AC: associated a series of the pulsating direct variable alternating volt volt Side 37



Experiment!!



- What's the highest frequency your eyes can notice?
 - frequency vs. amplitude..
- What is the frequency of 110V battery?



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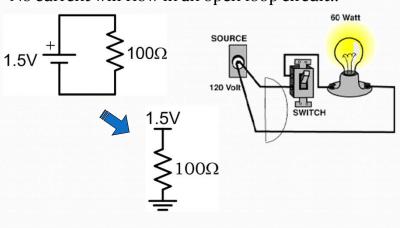
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Closed Loop Circuits

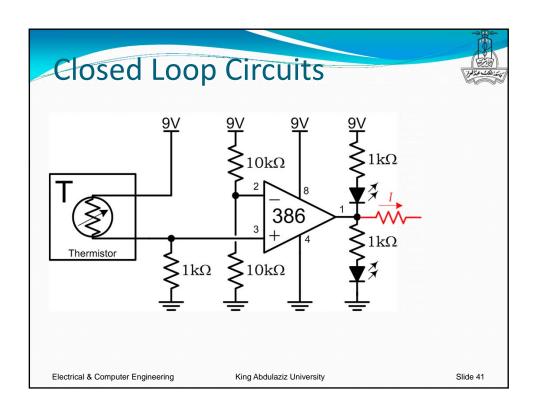


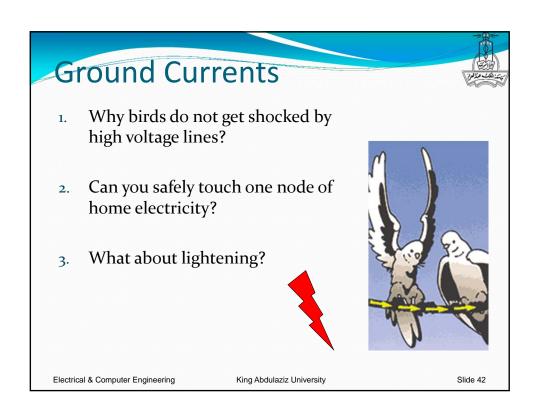
• No current will flow in an open loop circuit..



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Current



- Number of electrons passing every second
- Ampere (A) is a unit of current
 - $1 A = 6.28 \times 10^{18} \text{ electrons/sec}$
- At home: 6oA,10oA (2oA per phase)
- Electronics: 10mA

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AC Source



• Generally produces sine wave



$$\begin{array}{ll} v_s &= v_{\textit{peak}} \times \sin(\omega \times t) \\ v_s &= v_{\textit{peak}} \times \sin(2\pi \times f \times t) \end{array}$$

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RMS



- RMS: Root Mean Square
 - power is measured instead of peak voltage

$$V_{RMS} \equiv \sqrt{\frac{1}{T} \int_{0}^{T} v^{2}(t) \cdot dt}$$

• for sine or cosine signals, $v(t) = A \cos(2\pi t/T)$

$$V_{RMS} = \frac{v_{peak}}{\sqrt{2}}$$



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RMS



- If your meter reads 10V~
 - then $v_{peak} = 10*1.4 = 14V$
- If your peak voltage is $154V_p$
 - then the meter reads RMS = 154*0.7 = 110V~

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